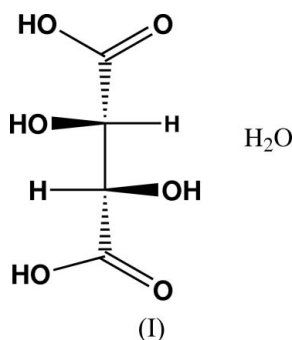


(2*S*,3*S*)-2,3-Dihydroxysuccinic acid monohydrateQing-Bao Song,<sup>a</sup> Ming-Yu Teng,<sup>a</sup>  
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The crystal structure of the title compound, C<sub>4</sub>H<sub>6</sub>O<sub>6</sub>·H<sub>2</sub>O, displays a two-dimensional network formed by O—H···O hydrogen bonds.Received 20 May 2006  
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## Comment

*D*-Tartaric acid is an important material in biochemistry and racemic resolution. The structure of racemic tartaric acid monohydrate was first determined and reported by Stern & Beevers (1950) and Parry (1951), and redetermined by Nie *et al.* (2001). The structure of *D*-tartaric acid tetrahydrate was determined by Okaya *et al.* (1966). However, *D*-tartaric acid monohydrate, (I), has not been reported until now.

In the crystal structure of (I), the molecules are linked by O—H···O hydrogen bonds (Table 1 and Fig. 2).

## Experimental

*D*-Tartaric acid (15.0 g, 0.10 mol) in water (9.0 ml) was heated to 343 K until the acid dissolved; the solution was then cooled to room temperature to give colourless prismatic crystals of (I) (5.5 g) after 5 d.

## Crystal data

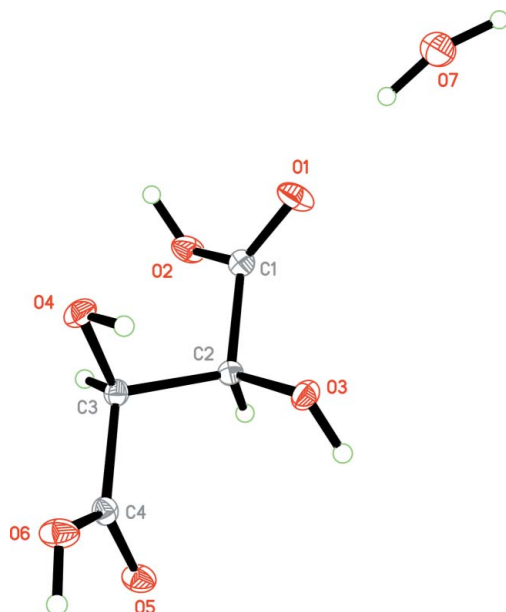
C<sub>4</sub>H<sub>6</sub>O<sub>6</sub>·H<sub>2</sub>O  
*M<sub>r</sub>* = 168.10  
Orthorhombic, *P*2<sub>1</sub>2<sub>1</sub>2<sub>1</sub>  
*a* = 7.6377 (7) Å  
*b* = 7.8268 (7) Å  
*c* = 11.0427 (10) Å  
*V* = 660.12 (10) Å<sup>3</sup>*Z* = 4  
*D<sub>x</sub>* = 1.691 Mg m<sup>-3</sup>  
Mo *K*α radiation  
*μ* = 0.17 mm<sup>-1</sup>  
*T* = 293 (2) K  
Prism, colourless  
0.51 × 0.44 × 0.43 mm

## Data collection

Bruker SMART CCD area-detector diffractometer  
*φ* and *ω* scans  
Absorption correction: multi-scan (*SADABS*; Bruker, 2000)  
*T<sub>min</sub>* = 0.779, *T<sub>max</sub>* = 1.000  
(expected range = 0.724–0.930)3853 measured reflections  
852 independent reflections  
818 reflections with *I* > 2σ(*I*)  
*R<sub>int</sub>* = 0.088  
*θ<sub>max</sub>* = 27.0°

## Key indicators

Single-crystal X-ray study  
*T* = 293 K  
Mean σ(*C*–*C*) = 0.004 Å  
*R* factor = 0.047  
*wR* factor = 0.126  
Data-to-parameter ratio = 6.6For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.



**Figure 1**  
The asymmetric unit of the title compound, showing 40% probability displacement ellipsoids and the atom-numbering scheme.

#### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.047$   
 $wR(F^2) = 0.126$   
 $S = 1.12$   
 852 reflections  
 129 parameters  
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0774P)^2 + 0.2665P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.001$   
 $\Delta\rho_{\max} = 0.32 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\min} = -0.32 \text{ e } \text{\AA}^{-3}$   
 Extinction correction: *SHELXL97*  
 Extinction coefficient: 1.13 (8)

**Table 1**

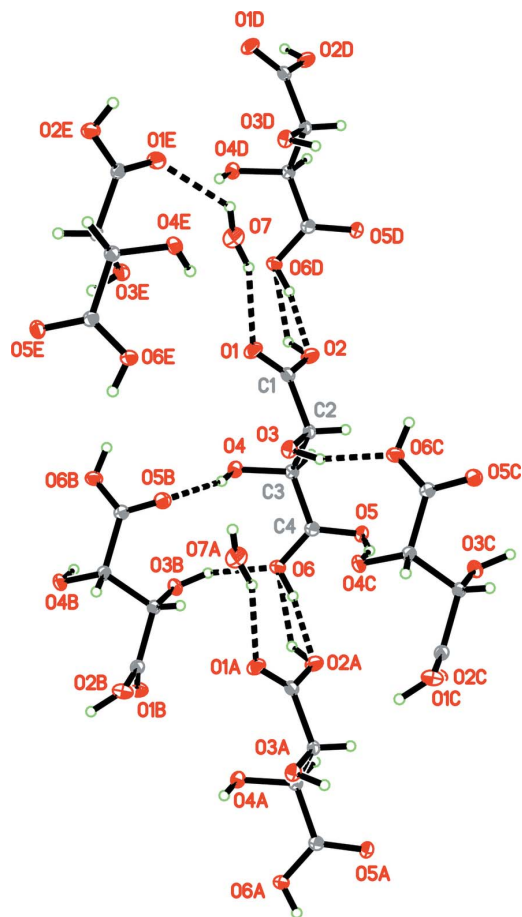
Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ ).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$O7-H7B\cdots O1$	0.92 (2)	2.04 (3)	2.882 (3)	151 (5)
$O7-H7A\cdots O2^i$	0.93 (2)	2.43 (4)	3.120 (4)	131 (4)
$O7-H7A\cdots O1^{ii}$	0.93 (2)	2.13 (4)	2.848 (3)	133 (4)
$O3-H5\cdots O6^{iii}$	0.90 (2)	1.88 (2)	2.737 (3)	158 (4)
$O4-H4\cdots O5^{iv}$	0.89 (2)	1.97 (2)	2.808 (3)	156 (4)
$O2-H1\cdots O6^v$	0.90 (2)	1.98 (7)	2.549 (3)	120 (6)
$O6-H6\cdots O2^{vi}$	0.82	1.74	2.549 (3)	169

Symmetry codes: (i)  $-x + \frac{3}{2}, -y + 1, z + \frac{1}{2}$ ; (ii)  $x + \frac{1}{2}, -y + \frac{3}{2}, -z + 1$ ; (iii)  $-x, y - \frac{1}{2}, -z + \frac{1}{2}$ ; (iv)  $-x, y + \frac{1}{2}, -z + \frac{1}{2}$ ; (v)  $x + 1, y, z$ ; (vi)  $x - 1, y, z$ .

Atom H6 was placed in a calculated position and allowed to ride on its parent atom at an O—H distance of 0.82  $\text{\AA}$ . Other H atoms were refined freely [O—H = 0.89 (2)–0.93 (2)  $\text{\AA}$  and C—H = 0.88 (4)–0.99 (4)  $\text{\AA}$ ].

Data collection: *SMART* (Bruker, 2000); cell refinement: *SAINT*; data reduction: *SAINT* (Bruker, 2000) and *SHELXTL* (Bruker,



**Figure 2**  
The O—H...O hydrogen bonds (dashed lines) in (I).

2000); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 2000); software used to prepare material for publication: *SHELXTL*.

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